

# 1 Environmental Contamination Following 2 the Grenfell Tower Fire

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7

## 8 **Abstract**

9 The Grenfell Tower fire in central London, started within a flat, engulfed the whole 24 storey  
10 building in flames, killed 72 people and spread toxic effluent via the plume and particulate  
11 deposits.

12 Soil samples from 6 locations up to 1,2 km from the Tower, together with semi-burnt fire debris  
13 and char samples, were collected 1 and 6 months after the fire. Additionally, dust samples and  
14 condensates were collected from a flat 160 m away from the Tower after 17 months. Samples  
15 were analysed for common potentially toxic components of fire effluents and synthetic vitreous  
16 fibres.

17 Samples collected within 140 m of the Tower showed, amongst other toxicants,  
18 polychlorinated dibenzo-p-dioxin concentrations 60 times greater than UK urban reference soil  
19 levels; benzene levels were 40 times greater; levels of 6 key polycyclic aromatic hydrocarbons  
20 (PAHs) were approximately 160 times greater. PAHs levels are approximately 20 times  
21 greater than those reported from nearby Hyde Park before the fire. To explain the presence

22 of these pyrogenic contaminants char and partially burnt debris were also collected and  
23 analysed. Benzene, PAHs, isocyanates and phosphorus flame retardants were found.  
24 Hydrogen cyanide and synthetic vitreous fibres were present in both soil and debris.

25 Particulate and pyrogenic contamination in the immediate vicinity is clearly evident, and may  
26 have leached out of fire debris, char and dust. Further analysis of the area around the Tower  
27 is necessary to understand potential health risks.

## 28 **1. Introduction**

### 29 **1.1 Harmful Effects of Fire Effluents**

30 UK National Fire Statistics (2018) show that the acute toxicity of fire effluents is the biggest  
31 short-term cause of death and injury from unwanted fires. Large fires produce smoke  
32 containing high concentrations of particulates and toxic gases such as, the asphyxiant gases,  
33 carbon monoxide (CO), hydrogen cyanide (HCN) and respiratory tract deep lung irritants. As  
34 the fire develops, the yields of all products of incomplete combustion including CO, HCN,  
35 organic compounds and soot increase - typically by factors of 10 to 50. Molecular toxicants  
36 bind to smoke particles (airborne soot and tarry droplets) allowing them to penetrate deep into  
37 the lung causing respiratory distress and pulmonary oedema (flooding of the lungs). This is  
38 closely followed by incapacitation and death, from few hours to several days or even years  
39 after exposure (Stec and Hull 2010; Stec 2017).

40 There have been surprisingly few reports of the long term consequences of unwanted fires.  
41 Persson and Simonson (1998) showed that in Sweden they contributed around 10% as much  
42 as transport-derived particulate emissions. Fires also release a rich cocktail of pollutants,  
43 many of them acutely or chronically toxic, including carcinogens such as semi and volatile  
44 organic compounds (SVOC/VOCs), PAHs, respiratory sensitizers such as isocyanates from  
45 some nitrogen-containing fuels, and persistent, bioaccumulative and toxic compounds such  
46 as polychloro- and polybromo dibenzo-p-dioxins and dibenzofurans (PCDD/Fs and PBDD/Fs)

47 and polychlorinated biphenyls (PCBs), formed by burning halogen containing fuels (McGee  
48 *et al.* 2003; Landrigan *et al.* 2004).

49 Benzene is a carcinogen in its own right (ATSDR 2018a). Other aromatic SVOC/VOCs are of  
50 particular toxicological significance as precursors of PAHs and carcinogens. Some PAHs,  
51 PCDD/Fs and PBDD/Fs (the most toxic is 2,3,7,8-tetrachlorodibenzodioxin (TCDD)) are also  
52 genotoxic and mutagenic (ATSDR 2018b). Benzo(a)pyrene (BaP) was initially identified as  
53 the most toxic PAH species, however more recent studies have identified 7,12-  
54 dimethylbenzo(a)anthracene as having a 20-fold higher toxic equivalence factor (TEF) than  
55 its parent compound and twice that of BaP (Andersson and Achten 2015). A study by Wang  
56 *et al.* (2009) showed that PAHs are transformed in the atmosphere or metabolically into  
57 hydroxy-PAHs, which are more genotoxic than the parental PAHs. These compounds have  
58 been linked to firefighter cancers through the analysis of their exposure (Stec 2018).

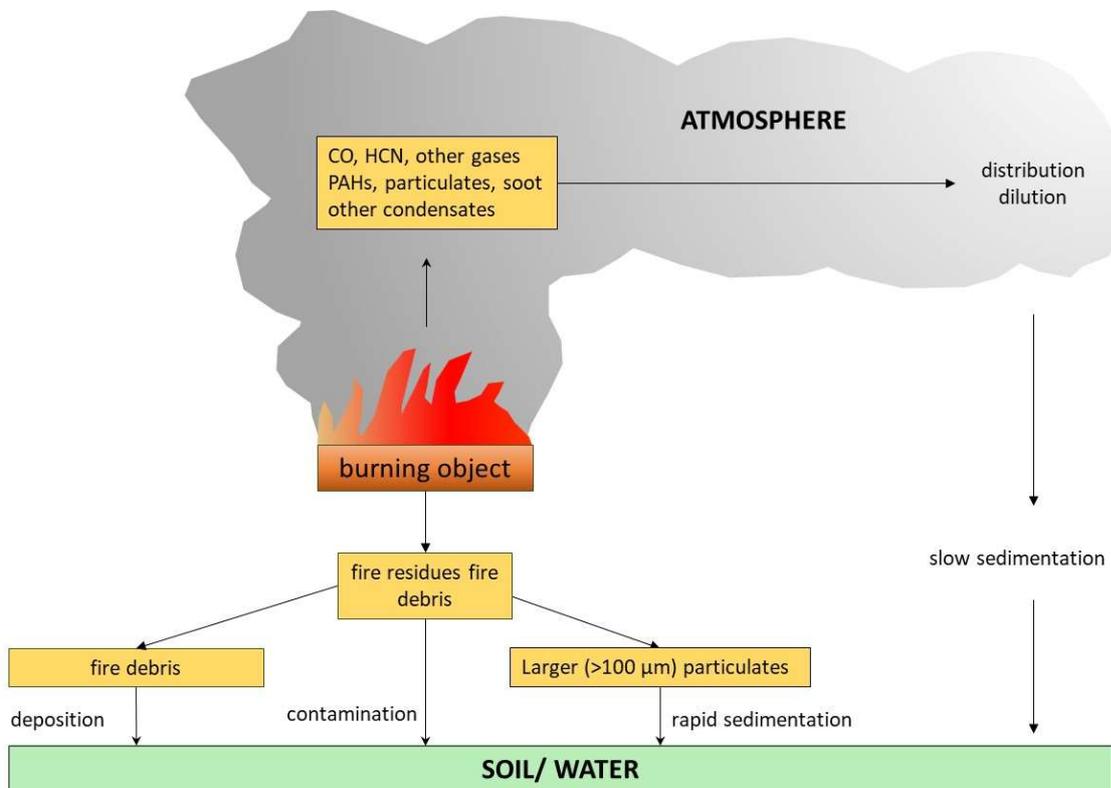
59 The study by Bengtström *et al.* (2016) showed that isocyanates have been positively identified  
60 in fire smoke and are widely used in the manufacture of flexible polyurethane (PU) foams for  
61 upholstered furniture and rigid PU or polyisocyanurate (PIR) foams for insulation in buildings.  
62 Isocyanates are respiratory sensitizers that can cause asthma attacks. They also trigger  
63 irritant and allergic forms of contact dermatitis (rashes, itching, swelling of extremities etc.)  
64 and less frequently hypersensitivity pneumonitis - an inflammation of the alveoli caused by  
65 inhaled isocyanate particles. A common decomposition product of isocyanates is methyl  
66 isocyanate (MIC) which also causes swelling of the lungs and breathing difficulties.

67 Studies by Lippmann (2014 and 2015) on the aftermath of the World Trade Centre showed  
68 that synthetic vitreous fibres (SVF) were one of the most significant health damaging  
69 contaminants after the fire. Inhalation exposure to airborne SVFs is a public health concern  
70 because like other particulate matter, fibres that are released in fires can be suspended in air  
71 (as dust or ash), inhaled and deposited in the lung (ATSDR 2018c). Lippmann (2014) identified  
72 the minimum critical fibre lengths for asbestosis (interstitial fibrosis), mesothelioma and lung  
73 cancer to be  $\sim 2 \mu\text{m}$ ,  $\sim 5 \mu\text{m}$  and  $\sim 15 \mu\text{m}$ , respectively. With regard to fibre diameter for

74 asbestosis and lung cancer, fibres with diameters  $>0.15 \mu\text{m}$  appear to be of predominant  
75 significance (as thinner fibres can be more readily cleared via the lymphatic system) whilst for  
76 mesothelioma (and other lesions of the mesothelium), fibre diameters  $<0.1 \mu\text{m}$  seem to be the  
77 most pathogenic.

## 78 1.2 Environmental Pathways

79 The interaction between a fire and its surroundings or environment proceeds via direct  
80 gaseous and particulate emissions to the atmosphere and localised deposition to soil and  
81 water. Subsequent dispersion and deposition of atmospheric emissions results in widespread,  
82 low level contamination of soil, ground and surface water, as shown Figure 1. Van Loon and  
83 Duffy (2000) reported that particles with diameters less than  $10 \mu\text{m}$  will have a deposition rate  
84 of around  $3 \text{ mm s}^{-1}$  and will tend to remain airborne, travelling with the smoke plume. Particles  
85 with diameters greater than  $100 \mu\text{m}$  will have a settling velocity of  $0.3 \text{ m s}^{-1}$  and are likely to  
86 be deposited close to the fire.



87  
88 **Figure 1.** Emission pathways from fires (adapted from ISO 26367-1 (2011)).

89 The degree to which fire species are partitioned between different phases (gaseous, aqueous,  
90 solid etc.) also depends on their physical characteristics and weather conditions (temperature,  
91 rain, wind speed etc.). For example, PAHs will agglomerate eventually leading to soot  
92 formation. The agglomerating species will initially travel as airborne particulates, but may grow  
93 large enough to sediment into water or soil, while CO will remain in the gas phase. Cyanide is  
94 released into air as a gas and to a lesser extent as particulate bound cyanides (ATSDR 2006).  
95 Cyanide can be transported over long distances before decomposition by reaction with  
96 hydroxyl radicals. In soil, HCN co-exists with alkali metal salts where it volatilises or degrades  
97 rapidly. Alternatively, HCN may be immobilised into metallo-cyanide complexes such as  
98 ferricyanides or ferrocyanides (ATSDR 2006). MIC will only persist in the atmosphere for a  
99 few hours to a few days, while in soil it will be broken down into other compounds upon contact  
100 with moisture (ATSDR 2014). PAHs and VOCs are comprised of species that partition  
101 differently according to their mass, with lighter species remaining primarily in the gaseous  
102 phase and heavier species tending to deposit on surface water or soil when absorbed on  
103 particulates (>2.5 µm) such as fly ash and soot (Van Loon and Duffy 2000). Humans can also  
104 be exposed to PAHs through inhalation or dermal contact with re-suspended soil and dust  
105 (Stec *et al.* 2018). While human-soil contact generally occurs outdoors, inhalation is also  
106 identified as a source of PAHs indoors, where people spent 80–93% of their time (WHO 2010).  
107 SVFs with smaller diameters become airborne more readily than fibres with larger diameters.  
108 SVFs remain unchanged in air, soil or sediment over long periods (Bernstein *et al.* 2005).

109 The UK's Public Health England (PHE) provides specialist advice on health including health  
110 advice on air quality, smoke exposure, asbestos, and the clean-up process (PHE 2018a). The  
111 data from the air quality monitoring in the area surrounding Grenfell Tower, since the start of  
112 the fire on 14 June, has shown that the risk to people's health from air pollution around the  
113 Grenfell Tower site was consistently low. Levels of gas particulate matter (PM<sub>10</sub>) remained low  
114 and monitoring results for dioxins, furans, PCBs and PAHs were broadly equivalent to  
115 background levels for London. No asbestos was reported as found, despite being present in

116 the Grenfell tower. There are no reports of contamination measurements being taken from the  
117 soil or water run-off. No measurements appear to have been carried out by UK's  
118 Environmental Agency or the local authority (the Royal Borough of Kensington and Chelsea  
119 (RBKC). RBKC are legally responsible for assessing and quantifying contaminated land within  
120 their community (PHE 2018b). The rationale for the current study was to address concerns  
121 from the Grenfell community related to the potential soil contamination and establish whether  
122 more detailed investigation is required.

## 123 **2. Materials and Methods**

124 Two char samples were collected from balconies 50 and 100 m from the Tower 1 month after  
125 the fire and analysed (Char1 and Char2). Based on the findings soil samples, together with  
126 fallen fire debris and more charred soot samples (Res and Char3) were collected 6 months  
127 after the fire at different distances from the Tower. Sampling was limited by locations where  
128 there was permission to collect soil and aimed to follow the direction of the prevailing wind at  
129 the time of the fire (South Easterly), with location shown in Figure 2a and wind on the day of  
130 fire Figure 2b (TimeandDate 2018). 17 months after the fire char from a balcony (Char4),  
131 indoor dust and a yellow oily deposit on a vertical fabric window blind (described by the  
132 occupier as "contaminated by the fire") were collected from a flat 160 m from the Tower. Table  
133 1 shows the details of the char and soil samples. A standard soil sample, Kettering loam soil,  
134 was obtained from Boughton Loam Ltd (containing clay 24%, silt 18%, sand 58%, organic  
135 content 6.72%). It is a preferred natural soil used as a standard in contamination analyses.

136 Quantitative analyses for PCDD/Fs, PAHs, benzene and metals were carried out on the char  
137 and soil samples. Gas chromatography-mass spectrometry (GC-MS) was used for  
138 SVOC/VOCs. Qualitative screening (thermogravimetric analysis coupled with gas phase  
139 Fourier Transform Infrared Spectroscopy, (TGA-FTIR)) was used to check for the presence of  
140 common fire effluents on all samples. Finally, the contaminated window blind was extracted  
141 and analysed for the presence of isocyanates, in order to characterise the yellow oily deposits.

142 [2.1 Sample Collection](#)

143 Soil samples (approximately 2 kg) were collected from the ground at depths of up to 200 mm.  
144 A fresh pair of gloves was used for each sample collection and the trowel was cleaned before  
145 and after each collection. The samples were stored in airtight 1 L dark glass jars covered in  
146 aluminium foil and kept at 4°C.

147 Approximately 60 pieces of what appeared to be char from insulation foam (the largest being  
148 300 mm in width and 460 mm in length, with an approximate density of 18 kg/m<sup>3</sup>) were  
149 collected from the ground within 90 m of the Tower. A semi-burnt piece of fire debris,  
150 recognisable as a sheet of insulation material (Res), was also found and collected. Samples  
151 were stored in dark polyethylene bags.

152 Char samples were also collected from three balconies (Char 1, 2 and 4) between 50 and  
153 160 m from the Tower. Dust samples were collected from five different locations within one  
154 apartment, 160 m from the Tower, and combined. Two pieces of the window blind, one with  
155 visible soot and yellow oily deposits and the other without, were also collected from the same  
156 apartment.

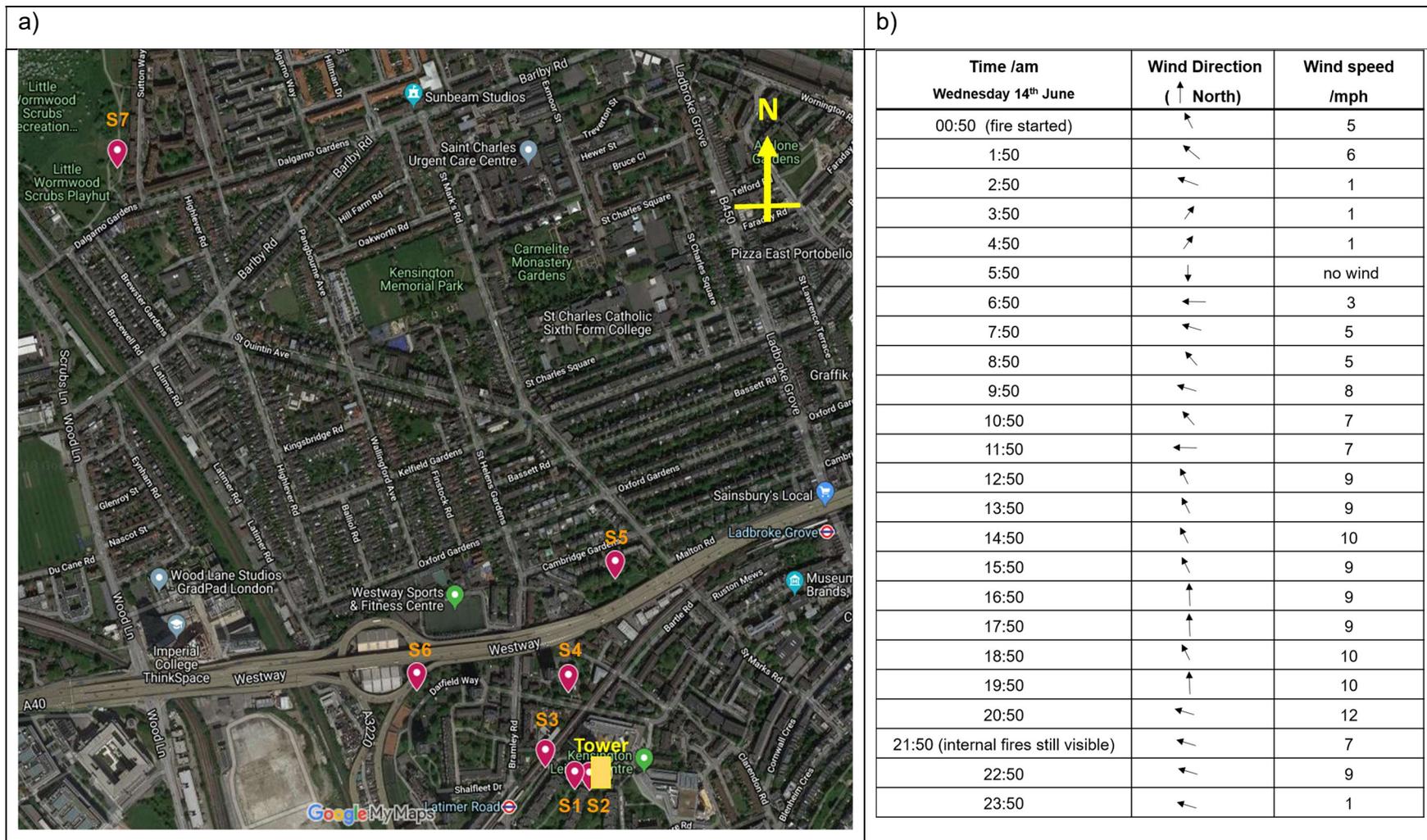
157 [2.2 Sample preparation](#)

158 Up to 5 g of each soil sample was then dried to a constant weight on a watch glass in an oven  
159 (VWR Dry-Line 115) at 60 °C to determine the moisture content, then sieved (5 mm) and  
160 ground to ensure a homogenous sample (the smell of fire smoke was observed for the soil  
161 samples 1 to 3). The moisture content, based on triplicate analyses, is reported in Table 1.  
162 Non-dried samples were used for TGA-FTIR analysis in order to avoid volatile losses.

163 **Table 1.** Sample descriptions, locations, moisture, pH and nitrogen content.

Sample Type	Abbreviation	Sample location* /(m)	Collection data**/ (months)	Average Moisture/ %	pH	Nitrogen content %
Kettering loam soil (control)	Ref			0.2	7.2	0.7
Soil	S1	27	6	34.9	5.4	1.9
	S2	50	6	35.7	6.1	1.7
	S3	90	6	26.2	8.5	1.2
	S4	142	6	24.2	4.2	1.2
	S5	316	6	23.6	6.7	1.2
	S6	332	6	14.9	6.9	1.2
	S7	1260	6	18.3	8.3	1.0
Fire debris	Res	90	6		6.4	UB: 2.1 B: 1.1
Char from balconies	Char1	50	1		-	6.1
	Char2	90	1		-	6.3
Charred samples from the ground	Char3	90	6		5.3	6.2
Char from balconies	Char4	160	17		5.5	4.9
Dust samples from the flat	Dust	160	17		-	-
Oily residue from window blind from the flat	WB	160	17		-	-

164 Notes: UB, Unburnt; B, Burnt; -, None detected; \*Sample location in respect to distance from Grenfell Tower; \*\*Data collection carried out months  
 165 after the fire; "Fire debris refers to semi-burnt insulation material.



166 **Figure 2.** Samples locations and meteorological reports of wind speed and direction during the fire (TimeandDate.com 2018).

167 2.2.1 Chemical analysis

168 All glassware was baked at 300 °C and rinsed with the appropriate solvent before use.  
169 Laboratory blanks were run alongside samples (intervals specified in individual sections  
170 below). All water was distilled. All samples were kept at 4 °C in a locked enclosure prior to  
171 analysis. All analyses were conducted in the analytical laboratories of the University of Central  
172 Lancashire except for the dioxins and furans which were quantified in a private UKAS  
173 accredited laboratory. The limits of detection (LOD) and limits of quantification (LOQ) for  
174 analysed fire effluents together with the dioxins and furans recoveries can be found in the  
175 supplementary material (Tables S1-S8).

176 2.2.1.1 pH

177 Approximately 20 g of each soil was mixed with 20 mL of deionised water and the water pH  
178 measured using a glass electrode in triplicate (Jenway 3540).

179 2.2.1.2 CHNS analysis

180 Approximately 2 mg of dried sample was placed into a tin capsule and run on a  
181 ThermoScientific Flash 2000 CHNS/O analyser (detection sensitivity within  $\pm 1\%$ ), in order to  
182 determine the presence of nitrogen. Each sample was analysed in triplicate with a blank run  
183 as part of the initial CHNS calibration daily. The instrument was calibrated with BBOT (2,5-Bis  
184 (5-tertbutylbenzoxazol-2-yl) thiopene) (Elemental Microanalysis, B2135) (6.51 N%, 72.53  
185 C%, 6.09 H%, 7.44 S%) using the K-factor calibration method. In place of laboratory blanks  
186 between samples, BBOT standard was run every 15 samples in order to check the response  
187 of the CHNS analyser.

188 2.2.1.3 ICP-OES screening

189 The method used was based on EPA 6010D (U.S. EPA 2014). 0.1 g of the sample was  
190 digested in 10 mL of concentrated nitric acid (Fisherbrand) in a microwave digester (Milestone  
191 Ethos EZ SR12) at 200 °C for 45 min. 0.1 mL of the digested sample was added to 9.9 mL of

192 water, which was then analysed by Inductively Coupled Plasma-Optical Emission  
193 Spectrometry (Thermo Scientific iCAP 7000 ICP-OES) for elemental composition. Samples  
194 were run in quadruplicate with each individual sample tested three times for consistency. The  
195 RSD for all final results was less than 5%. Blanks prepared from digested acid were run after  
196 every fifteen samples, and were all below the limits of detection (LOD) for all elements  
197 analysed. The LOD and LOQ were calculated as three and ten times the standard deviation  
198 from the analysis of the standards and the blanks (Table S1). The standards used for  
199 comparison were the TraceCERT® 1000 mg/L P in water and the multi-element standard 5  
200 TraceCERT® in 10% nitric acid (Sigma Aldrich).

## 201 2.2.2 Fire effluent analysis

### 202 2.2.2.1 Benzene quantification

203 3.5 g of soil sample, or 1 g of residue or char material, was added to 3 mL of a 3:1  
204 hexane:acetone mixture (Sigma Aldrich) and the sample was sonicated (Sonicor SC52-H) for  
205 40 min. 1.5 mL of the solution was extracted, centrifuged at 15 000 rpm for 30 min (Sanyo  
206 MSE Micro Centaur MSB010.CX2.5) and 2 µL of the extracted solution was injected into the  
207 GC-MS (Agilent HP 6890 coupled to Agilent MS 5973) with the software: Agilent MSD  
208 Chemstation version F.01.00.1903). All analyses were run in quadruplicate. Laboratory blanks  
209 run every ten samples. The chromatograms used for analysis were blank subtracted. The LOD  
210 was calculated using three times the signal to noise ratio of the analyte, while the LOQ was  
211 calculated using ten times the signal to noise ratio. The LOD and LOQ were 0.11 and  
212 0.54 ppm respectively.

### 213 2.2.2.2 PAHs and phosphorus flame retardants

214 5 g of soil was passed through a 5 mm sieve and added to 40 mL of a 1:1 ratio of  
215 dichloromethane:acetone mixture (Sigma Aldrich) and sonicated (Sonicor SC52-H) for 2 h.  
216 For the next 6 h the samples were sonicated for 10 min each hour. Once sonication was  
217 complete, 2 mL of the liquid was centrifuged at 15 000 rpm for 30 min (Sanyo MSE Micro

218 Centaur MSB010.CX2.5) and 2 µL of the centrifuged extracted solution was injected into the  
219 GC-MS (Agilent HP6890 coupled to Agilent MS 5973). Approximately 1 g of residue or char  
220 sample was added to 3 mL of a 3:1 hexane:acetone mixture (Sigma Aldrich) and the samples  
221 sonicated for 40 min. 1.5 mL of the solution was extracted, centrifuged at 15 000 rpm for 30  
222 min and 2 µL of the extracted solvent was injected into the GC-MS. Each analysis was  
223 repeated five times.

224 Quantification of PAHs was based on the method described by Guerin (1999). Analysis was  
225 performed for PAHs using a HP 6890 Series GC System equipped with a 5973 Mass Selective  
226 Detector (Hewlett Packard) and a TraceGOLD TG-5MS column with the dimensions  
227 30 m x 0.25 mm x 0.1 µm (Thermo Scientific). An injection volume of 2 µL was used with a  
228 splitless injection with a flow rate of 35 mL/min. Samples were analysed in SIM mode. The GC  
229 was set to a start temperature of 100 °C with a 2 min hold, then with 8 °C/min to 210 °C, with  
230 2 °C/min to 280 °C and was held at 280 °C for 3 min. The results were then processed using  
231 Agilent MSD Chemstation version F.01.00.1903. Calibration standards were obtained from  
232 Sigma Aldrich. Laboratory blanks were analysed with every ten samples. The PAHs were  
233 quantified using external standard calibrations. The LOD was based on three times the signal-  
234 to-noise ratio of each analyte (related to the 5g samples) while the LOQ was based on ten  
235 times the signal-to-noise ratio, as shown in Table S2. Responses below the LOQ were not  
236 included in this analysis. The average blank levels were below the LOD for all PAHs.

#### 237 2.2.2.3 Polychlorinated dibenzo-p-dioxin and polychlorinated dibenzofuran analysis

238 Quantification of PCDD/Fs was based on EPA1613 (US EPA 1994). The analysis was  
239 undertaken in a UKAS accredited laboratory, approved to quantify dioxins. This includes a  
240 spiked sample and a reference material analysed alongside the samples on a weekly basis.  
241 The LOD and recoveries for each sample are shown in the supplementary document  
242 (Table S3-S8). Approximately 1 g of sample was fortified with <sup>13</sup>C dioxin/furan standards  
243 (Wellington) at 1 ng for each congener. Samples were then extracted in a Soxhlet apparatus  
244 in toluene (Rathburn), for 16 h. Samples were solvent extracted with hexane (Rathburn)

245 followed by rotary evaporation (Buchi 310). Sample clean-up was achieved using a Miura  
246 system (GO-HT) with silica, alumina and carbon columns - eluting with hexane and then  
247 toluene. The final volume of samples was reduced to 20  $\mu$ L and fortified with recovery  
248 standards ((Wellington). Analysis was carried out by using 60 m DB5 capillary column (Agilent)  
249 and Waters NT Ultima high resolution mass spectrometer operating at 10000 resolving power,  
250 which was quantified against isotope dilution calibration curve. Analysis was done using  
251 Masslynx Software.

#### 252 2.2.2.4 Isocyanate analysis

253 0.5 g of each sample was added to 1 mL of a 0.01 M di-n-butylamine in toluene solution (Sigma  
254 Aldrich) and sonicated (Sonicor SC52-H) for 30 min. Post-sonication 500  $\mu$ L of a 500 ng/mL  
255 standard solution was added to 500  $\mu$ L of extract and the samples were evaporated to dryness  
256 under a stream of nitrogen (Energas). 0.5 mL of UPLCMS grade acetonitrile (Fisherbrand)  
257 was then added to the vial which was gently swirled for 30 s before the solution was transferred  
258 to the UPLCMS system (Bengtström *et al.* 2018). The results were compared to a standard  
259 solution containing a number of isocyanates (Sigma Aldrich, DBA Isocyanate Standard  
260 Mixture). The UPLCMS was purged before testing with the UPLCMS grade solvents, and three  
261 blanks were run immediately prior to the samples. Due to the low quantity of the samples, they  
262 were treated as qualitative samples and the MS spectra compared to spectra obtained from a  
263 purchased calibration standard mixture used as a reference. The detailed analytical settings  
264 are presented in Table 2.

265

266 **Table 2.** UPLCMS settings for the isocyanate analysis.

<b>UPLCMS settings</b>	
Equipment specifications	Waters Acquity UPLC coupled to a Waters TQD MS Software: Mass lynx version V4.1 SCN714
Column	Waters BEH C18 column (130Å, 1.7 µm, 2.1 mm X 50 mm)
Mobile Phases	A) 95:5 v/v water (VWR, 83645.320) and acetonitrile (VWR, 84865.260) and 0.05% formic acid (VWR, 20318.297) B) Acetonitrile and 0.05% formic acid
Flow rate	0.25 ml/min
Gradient program (time/min - %A)	Start – 40, 5 – 2, 6 – 40, 9 – 40
Total program length (min)	10
Column temperature (°C)	40

267

268 **2.2.2.5 TGA-FTIR analysis**

269 Thermogravimetric analysis coupled to gas phase Fourier Transform Infrared Spectroscopy  
 270 (TGA-FTIR) was used to qualitatively assess the presence of other fire effluents as a function  
 271 of sample's temperature. Additional to analyses of the soil samples, cyanide may be present  
 272 in soil as ferricyanide ( $\text{Fe}(\text{CN})_6^{3-}$ ) or ferrocyanide ( $\text{Fe}(\text{CN})_6^{4-}$ ), both analysed by TGA-FTIR in  
 273 order to determine the HCN release temperatures from the iron complexes. Samples were  
 274 analysed in a Mettler Toledo TGA/DSC2 connected directly to a Thermo Nicolet iS50 FTIR via  
 275 a heated line held at 250 °C. 100 mg of soil and 5 mg of residue were heated at 10 °C/min  
 276 from ambient to 700 °C in nitrogen. The FTIR was set to a resolution of 1  $\text{cm}^{-1}$  with a DGTS  
 277 detector and a scan rate of 10 averaged spectra in a 10 cm pathlength gas cell. The FTIR  
 278 spectra were compared with the HR Nicolet Vapour Phase and TGA Vapour Phase libraries  
 279 using OMNIC 9.3.32 software. The wavelength range used for HCN profiles was 3278-3292  
 280  $\text{cm}^{-1}$ . Each sample was analysed in triplicate in both air and nitrogen. The sample crucibles  
 281 were cleaned after each run. TGA-FTIR instrument was cleaned by and an isotherm run after  
 282 five tests. In addition, the gas phase FTIR was run using automatic atmospheric suppression.

283 **2.2.2.6 Synthetic vitreous fibre analysis**

284 SVFs were found in samples of soil, char and residue by manual searching and optical  
285 microscopy (Nikon Eclipse E200). Composition of SVF was estimated using a Jeol JCM-6000  
286 with Scanning Electron Microscope (SEM) using BED-C COMPO with an integrated JED-2300  
287 Energy Dispersive X-Ray Spectroscopy module (EDX). The software programs used for  
288 analyses were JCM-6000 Plus version 1.4.0 and Analysis Station version 3.8.0.59. The fibres  
289 were prepared on carbon stickers which were placed on SEM stubs for the analyses.

290 **3. Results**

291 **3.1 ICP-OES analysis**

292 Aluminium, zinc, copper, lead and other metals were present in soil within UK Environment  
293 Agency baseline pollutant levels in soil (EA 2007a). Phosphorus, occurring naturally in the  
294 soil, was present at higher levels for soils S1-S3 collected near the Tower (within the range of  
295 140-170 mg/kg) than for S4 to S7 with values between 85 and 35 mg/kg, respectively.

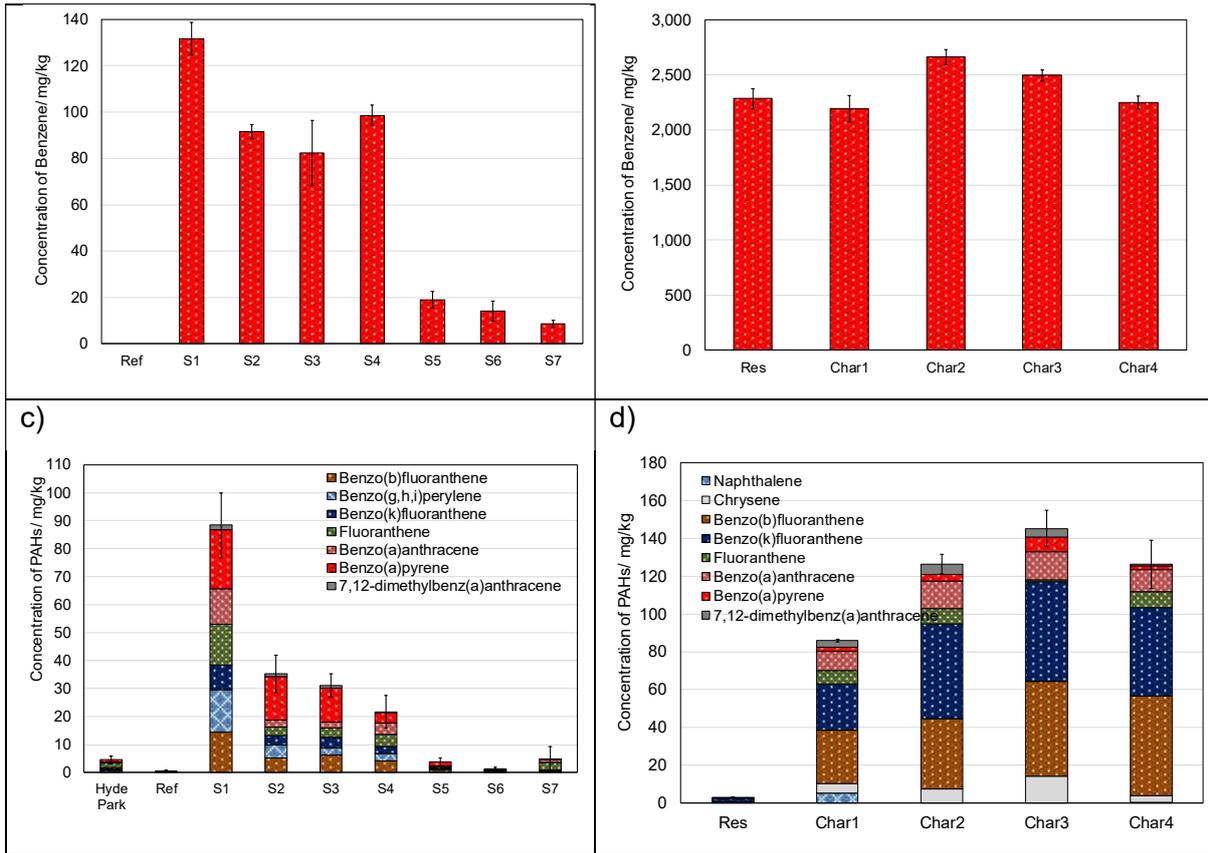
296 **3.2 Benzene analysis**

297 Data in Figure 3 shows elevated benzene concentrations for the first 4 soil samples when  
298 compared to the residential soils (EA 2007b). Benzene concentrations decrease with  
299 increasing distance from the Tower.

300 **3.3 PAHs quantification**

301 The sum of the six main PAH concentrations (BaP, fluoranthene, benzo(a)anthracene,  
302 benzo(b)fluoranthene, benzo(k)fluoranthene, and benzo(ghi)perylene) in London's urban  
303 soils, sampled in nearby Hyde Park (3.9 km away from the Tower), is estimated at 4512 µg/kg  
304 (as an average from three locations) (EA 2007b). Similarly to benzene, PAH concentrations -  
305 Figure 3 (c and d), show decrease with increasing distance from the Tower.

a)	b)
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306 **Figure 3.** Benzene (a and b) and PAHs (c and d) concentrations showing errors bar for total  
 307 concentrations in soil samples and fire debris in mg/kg.

308 **3.4 PCDDs and PCDFs analysis**

309 Table 3 shows median soil concentrations of different PCDDs and PCDFs found in soil  
 310 samples around Grenfell Tower compared to values in urban UK locations and nearby Hyde  
 311 Park (EA 2009a). PCDD/Fs concentrations are higher for the soil samples collected closer to  
 312 the Tower, appearing to peak around 100 m from it.

313 **Table 3.** PCDD/Fs levels in soil samples and Char3.

	UK urban/ ng/kg	Hyde Park/ ng/kg	Ref/ ng/kg	S1/ ng/kg	S2/ ng/kg	S3/ ng/kg	S4/ ng/kg	Char3/ ng/kg
<b>Dioxins</b>								
2,3,7,8-TCDD	0.4	0.5	-	1.05	1.3	1.6	-	-
1,2,3,7,8-PeCDD	1.4	1.6	-	8.7	6.7	9.2	1.3	1.8
1,2,3,4,7,8-HxCDD	1.6	1.5	0.3	14.9	14	16.8	1.3	1.5
1,2,3,6,7,8-HxCDD	2.9	3.5	0.7	39.8	36.8	45.7	2.9	2.9
1,2,3,7,8,9-HxCDD	2.6	3.0	0.5	30.2	31.4	34.9	2.5	2.5
1,2,3,4,6,7,8- HpCDD	25.2	27.5	7.3	889	965	1120	54.8	15.7
OCDD	104	88.8	25.3	6450	7370	8730	356	27.8
<b>Furans</b>								
2,3,7,8-TCDF	2.5	4.4	-	8.3	5.5	11	3.8	22.5
1,2,3,7,8-PeCDF	2.8	3.8	-	5.7	3.0	3.6	2.4	8.8
2,3,4,7,8-PeCDF	4.1	6.9	-	8.4	6.6	5.2	3.2	17.3
1,2,3,4,7,8-HxCDF	3.6	3.3	0.7	8.7	6.7	7.4	3.5	11.9
1,2,3,6,7,8-HxCDF	2.6	2.5	0.4	6.4	6.2	4.5	2.4	14.5
2,3,4,6,7,8-HxCDF	3.8	4.5	0.6	9.8	11.2	9.0	3.1	19.3
1,2,3,7,8,9-HxCDF	1.1	0.9	-	3.1	2.7	0.7	-	5.2
1,2,3,4,6,7,8- HpCDF	24.2	23.6	3.7	83.6	103	116	21.5	52.0
1,2,3,4,7,8,9- HpCDF	1.4	1.0	-	8.9	9.5	12.5	1.8	7.6
OCDF	26.0	19.4	3.7	202	222	239	23.6	19.6
∑PCDDs	138	126.5	34.1	7434	8425	9958	416	52.2
∑PCDFs	72.1	70.3	9.1	345	376	409	65.3	179
<b>Total PCDDs and PCDFs</b>	<b>210</b>	<b>197</b>	<b>43</b>	<b>7779</b>	<b>8802</b>	<b>10367</b>	<b>482</b>	<b>231</b>
<b>TEQ-WHO (mammals)</b>	<b>6.48</b>	<b>8.65</b>	<b>1.39*</b>	<b>36.4*</b>	<b>34.6*</b>	<b>40.6*</b>	<b>5.63*</b>	<b>16.4*</b>

314 Notes: ∑, sum of; -, None detected; \* concentration of non-detected congeners at detection  
315 limit

### 316 3.5 Phosphorus flame retardants analysis

317 Tris(chloroisopropyl) phosphate (TCPP), tris(2-ethylhexyl) phosphate (TEHP) and tricresyl  
318 phosphate (TCP) were identified in samples S1 and S2, fire debris and Char3. These are  
319 commonly used in insulation foam and upholstered furniture foam and do not occur naturally  
320 in the soil (Hewitt *et al.* 2017).

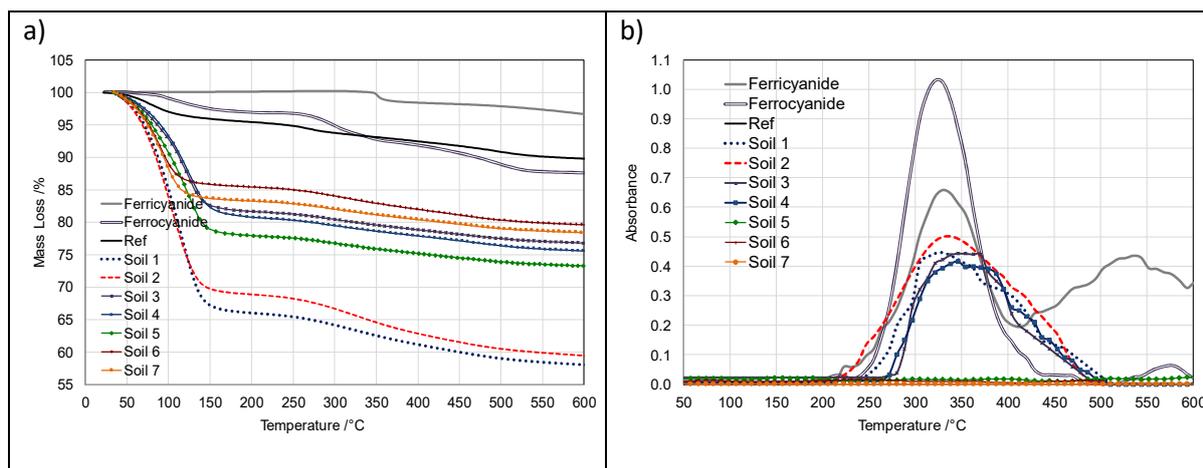
### 321 3.6 Isocyanates analysis

322 MIC, isocyanic acid, ethyl isocyanate and propyl isocyanate were identified in the yellow  
323 deposit on the window blind.

### 324 3.7 TGA-FTIR analysis

325 TGA shows the mass loss as a function of temperature when the sample is heated in an inert  
326 atmosphere (nitrogen) while FTIR allows identification of volatiles as a function of temperature.  
327 Thus, the species observed at elevated temperatures may have been trapped or otherwise  
328 adsorbed particularly if they were released below 150 °C. At higher temperatures they are  
329 more likely to be decomposition products.

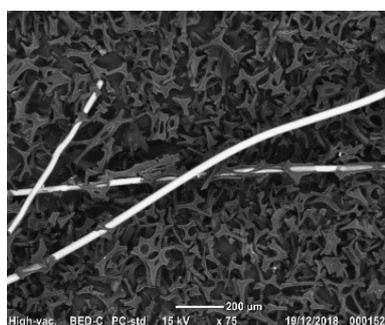
330 There are broad similarities in terms of released volatiles, shown in Figure 4a. Initial TGA  
331 mass losses, up to around 120 °C, are largely assigned to water release followed by other  
332 volatiles. HCN and alkyl cyanides were evolved from over a temperature range of 210 °C.  
333 HCN release from soil samples (S1, S2, S4 and S5), fire debris and char collected from  
334 individuals balconies is observed from 280 to 310 °C, reaching a peak of around 350 °C.  
335 Ferrocyanide releases HCN between 250 and 450 °C, while ferricyanide have two distinguish  
336 HCN release maxima at 330 °C and 540 °C. Figure 4b shows the HCN profile of the iron  
337 cyanide complexes alongside S1 to S7.



338 **Figure 4.** TGA mass loss curves of soil, Res and Char3 samples (a) and HCN release  
 339 profiles for the soil samples (b).

### 340 3.8 Synthetic Vitreous Fibres analysis

341 SVFs were identified and isolated from soil samples S1 and S2 and were found attached to  
 342 fallen debris and char samples. In order to identify the possible origins of the SVFs, samples  
 343 of commercially available polyisocyanurate (PIR), phenolic foam (PhF) and stone wool (SW)  
 344 were used as reference samples. The PIR sample had two layers of glass wool embedded at  
 345 depths of approximately 25 and 55 mm within the foam. The phenolic foam sample had layers  
 346 of glass wool embedded into the outer surface layers inside the foil covering. The foil covering  
 347 was also reinforced with glass fibres. Reference samples were of the same manufacturer and  
 348 product names as those reported to have been used on the Tower refurbishment (Grenfell  
 349 Tower Inquiry 2018). An example of the SVF attached to the foam residue can be seen in  
 350 Figure 5 and elemental composition in Table 4. Samples were run in quadruplicate.



351  
 352 **Figure 5.** Picture of SVF in Char3.

353 **Table 4.** Elemental analysis of the fibres.

	<b>Abbr.</b>	<b>Na<sub>2</sub>O</b>	<b>MgO</b>	<b>Al<sub>2</sub>O<sub>3</sub></b>	<b>SiO<sub>2</sub></b>	<b>P<sub>2</sub>O<sub>5</sub></b>	<b>SO<sub>3</sub></b>	<b>K<sub>2</sub>O</b>	<b>CaO</b>	<b>TiO<sub>2</sub></b>	<b>FeO</b>	<b>Diameter range /μm</b>
Reference Materials	SW	3.7	9.7	19.4	39.5	0.7	1.6	0.9	17.8	1.8	6.0	8.4 – 11.1
	PIR	19.8	3.8	2.5	70.0		0.3	0.4	3.6			24.5-29.9
	PhF	1.7	3.4	17.0	59.7	2.5	8.2	0.8	15.3			9.7-14.6
Samples	Res	3.7	2.3	15.5	53.5		7.0	0.9	17.9	3.7	2.3	12.1 - 14.4
	Char2	15.2	4.3	2.2	72.3	1.0	0.4	0.5	4.8	15.2	4.3	28.8-31.7
	Char3	12.8	3.9	2.3	74.8	1.0	0.2	0.5	4.7	12.8	3.9	25.1-28.8
	S1	17.8	3.4	2.5	71.3		0.2	0.5	4.3	17.8	3.4	24.5 - 26.1
	S2	1.2	2.3	15.3	57.4	0.8	0.9	0.3	22.1	1.2	2.3	12.0 - 12.5

354 Note: Abbr, Abbreviation

#### 355 4. Discussion and Conclusions

356 Soil guideline values provide a reliable baseline against which intensive local surveys and  
357 future national surveys can be assessed (EA 2009b). If representative soil concentrations are  
358 above the baseline values then further investigation is required to determine whether the  
359 substances pose a risk and to determine the scale and urgency of further action. The greater  
360 the exceedance of the assessment level, the greater the likelihood that the substance will pose  
361 a risk to human health and/or the environment (EA 2009b).

362 Soil guideline values for benzene for residential land are between 0.87 – 3.3 mg/kg and 95  
363 mg/kg for commercial land use where contamination is expected (i.e. petrochemical and  
364 petroleum refining industries) (PHE 2014; EA 2009c). Values are based on a consideration of  
365 the oral, dermal and inhalation routes of exposure. Benzene is typically found in petrol and in  
366 vehicle emissions, therefore elevated levels may be expected at roadside locations. Data in  
367 Figure 3 shows benzene concentrations exceeded these values by factors between 25 and  
368 40 for the four soil samples taken closest to the Tower when compared to the residential soils.

369 The British Geological Survey defined the normal background concentrations for BaP in  
370 England to be 3.6 mg/kg in urban areas and 0.5 mg/kg in all other areas. Guidance levels for  
371 BaP are set at 5.0 to 5.3 mg/kg for residential land and up to 77 mg/kg for commercial land  
372 (EA 2007b; PHE 2018c). BaP concentrations obtained from the first three soil samples exceed  
373 residential values (33, 24 and 17 mg/kg respectively). For soil S5 the value is 2.0 mg/kg, with  
374 S6 and S7 values at 0.3 mg/kg - showing the localised distribution of the contamination. The  
375 total sum of 6 PAH concentrations (S1), 45 m away from the Tower, is approximately 20 times  
376 higher than that reported in Hyde Park (or approximately 160 times greater than the reference  
377 soil). S2 to S4 exceeded these reference values by factors between 40 and 60. S5 to S7 are  
378 comparable to the reference soils. PCDD levels are around a factor of 70 greater than those  
379 collected in Hyde Park or a factor of 60 greater than the UK urban reference soil values. S4

380 contains lower concentrations than S1 to S3, but these are still three times higher than the UK  
381 urban or Hyde Park concentrations.

382 Seventeen 2,3,7,8-substituted PCDD and PCDF congeners (Table 3) and 15 PAHs (BaP,  
383 benzo(a)anthracene, benzo(b)fluoranthene, benzo(k)fluoranthene, indeno(1,2,3-cd)pyrene,  
384 chrysene, acenaphthene, acenaphthylene, anthracene, benzo(g,h,i)perylene, fluoranthene,  
385 fluorene, phenanthrene, pyrene and naphthalene) with their respective TEFs were used to  
386 conduct non-cancer and cancer risks assessment using 2,3,7,8 – TeCDD and BaP toxicity  
387 equivalence (PCDD/F WHO-TEQ and PAHs WHO-TEQ) (Van den Berg *et al.* 2013; U.S. EPA  
388 2001; 2009; 2018). Results, as Non-Cancer Hazard Quotient (HQ of 1 for non-carcinogens)  
389 and Cancer Risk (CR corresponding to a  $10^{-6}$  risk level for carcinogens) are presented in Table  
390 5. Reference doses, slope factors and other parameters for estimating human non-cancer and  
391 cancer risks were taken from Regional Screening Levels Tables and EPA equations (U.S.  
392 EPA 1989; 1991; 2001; 2009; 2018). In this study, the body weight was chosen 70 kg for  
393 adults and 15 kg for children. Exposure duration of 25 years for adults and 6 years for children  
394 was chosen with ingestion rates of soil 30 and 15 mg/day, respectively. Exposure frequency  
395 was assumed to be 50 days/year and exposure time of 1 hr/day for inhalation, ingestion and  
396 dermal pathway. Surface area of skin that contacts the soil was taken as 1500 and 500 cm<sup>2</sup>  
397 for adults and children, respectively (EA 2008). A life time average of 60 years was taken to  
398 calculate the average time exposure for carcinogenic chemical exposure.

399 The Hazard Quotient (HQ) together with the lifetime cancer risk was calculated and is  
400 presented in Table 5.  $HQ \leq 1$  indicates no adverse health effects, whereas  $HQ > 1$  indicates  
401 likely adverse health effects (NYS DOH 2007). An estimated increased excess lifetime cancer  
402 risk is not a specific estimate of expected cancers (when values exceed the unity). Rather, it  
403 is a plausible upper bound estimate of the probability that a person may develop cancer  
404 sometime in his or her lifetime following exposure to that contaminant (Van den Berg *et al.*  
405 1998).

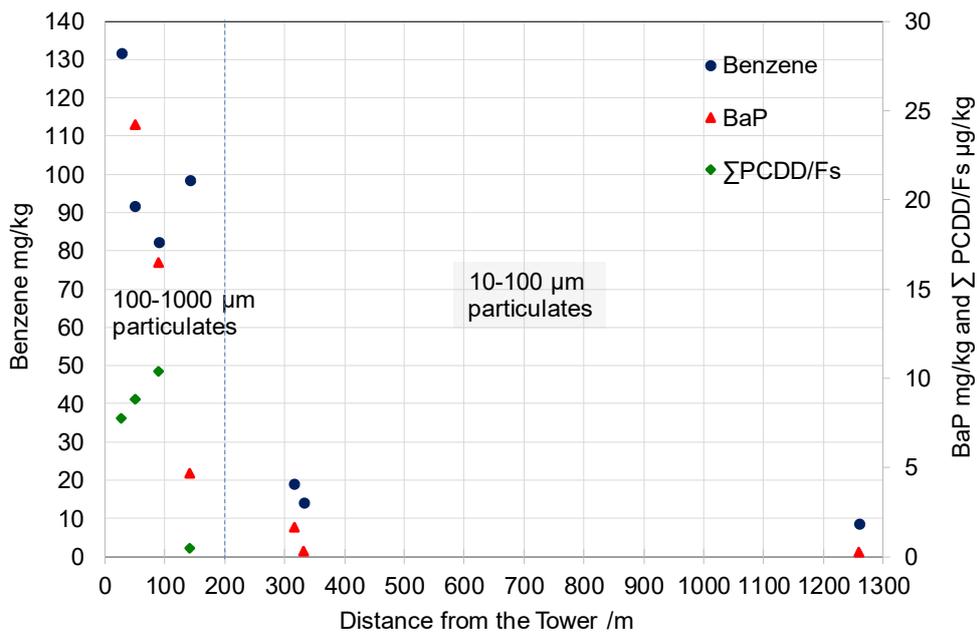
406 **Table 5** Hazard quotient and cancer risk from exposure to residue, soil and char samples.

	Hazard quotient (HQ) of adult			Hazard quotient (HQ) of children			Cancer risk of human ( $10^{-6}$ )		
	Ingestion	Dermal contact	Inhalation	Ingestion	Dermal contact	Inhalation	Ingestion	Dermal contact	Inhalation
<b>Dioxins/Furans</b>									
<b>UK urban</b>	6.43E-05	3.86E-05	3.86E-09	6.01E-04	6.01E-05	3.86E-09	5.86E-03	5.02E+06	5.87E-12
<b>Hyde Park</b>	6.02E-05	3.61E-05	1.45E-16	5.62E-04	5.62E-05	3.62E-09	5.48E-03	4.70E+06	1.45E-10
<b>Ref</b>	1.32E-05	7.93E-06	3.18E-17	1.23E-04	1.23E-05	7.94E-10	1.20E-03	1.03E+06	3.18E-11
<b>S1</b>	2.38E-03	1.43E-03	5.72E-15	2.22E-02	2.22E-03	1.43E-07	2.17E-01	<b>1.86E+08</b>	5.72E-09
<b>S2</b>	2.69E-03	1.62E-03	6.47E-15	2.51E-02	2.51E-03	1.62E-07	2.45E-01	<b>2.10E+08</b>	6.47E-09
<b>S3</b>	3.17E-03	1.90E-03	7.62E-15	2.96E-02	2.96E-03	1.91E-07	2.89E-01	<b>2.48E+08</b>	7.62E-09
<b>S4</b>	1.48E-04	8.89E-05	3.56E-16	1.38E-03	1.38E-04	8.90E-09	1.35E-02	<b>1.16E+07</b>	3.56E-10
<b>Char3</b>	7.07E-05	4.24E-05	1.70E-16	6.60E-04	6.60E-05	4.24E-09	6.43E-03	5.51E+06	1.70E-10
<b>Polycyclic Aromatic Hydrocarbons</b>									
<b>Ref</b>	9.55E-05	3.72E-02	2.46E-03	8.91E-04	5.79E-02	2.46E-03	2.86E-02	9.68E+03	2.47E-11
<b>S1</b>	1.95E-02	<b>7.61E+00</b>	5.02E-01	1.82E-01	<b>1.18E+01</b>	5.02E-01	5.85E+00	<b>1.98E+06</b>	5.05E-09
<b>S2</b>	1.23E-02	<b>4.80E+00</b>	3.17E-01	1.15E-01	<b>7.47E+00</b>	3.17E-01	3.69E+00	<b>1.25E+06</b>	3.19E-09
<b>S3</b>	9.76E-03	<b>3.81E+00</b>	2.51E-01	9.11E-02	<b>5.92E+00</b>	2.51E-01	2.93E+00	<b>9.89E+05</b>	2.53E-09
<b>S4</b>	3.47E-03	<b>1.35E+00</b>	8.92E-02	3.24E-02	<b>2.10E+00</b>	8.92E-02	1.04E+00	3.51E+05	8.98E-10
<b>S5</b>	1.03E-03	4.00E-01	2.64E-02	9.57E-03	6.22E-01	2.64E-02	3.08E-01	1.04E+05	2.66E-10
<b>S6</b>	2.42E-04	9.45E-02	6.23E-03	2.26E-03	1.47E-01	6.23E-03	7.27E-02	2.46E+04	6.28E-11
<b>S7</b>	2.92E-04	1.14E-01	7.51E-03	2.72E-03	1.77E-01	7.51E-03	8.75E-02	2.96E+04	7.56E-11
<b>Res</b>	1.09E-03	4.26E-01	2.81E-02	1.02E-02	6.63E-01	2.81E-02	3.28E-01	1.11E+05	2.83E-10
<b>Char1</b>	6.20E-03	<b>2.42E+00</b>	1.59E-01	5.78E-02	<b>3.76E+00</b>	1.59E-01	1.86E+00	6.28E+05	1.61E-09
<b>Char2</b>	9.89E-03	<b>3.86E+00</b>	2.55E-01	9.24E-02	<b>6.00E+00</b>	2.55E-01	2.97E+00	<b>1.00E+06</b>	2.56E-09
<b>Char3</b>	1.41E-02	<b>5.51E+00</b>	3.63E-01	1.32E-01	<b>8.57E+00</b>	3.63E-01	4.24E+00	<b>1.43E+06</b>	3.66E-09
<b>Char4</b>	9.29E-03	<b>3.62E+00</b>	2.39E-01	8.67E-02	<b>5.63E+00</b>	2.39E-01	2.79E+00	9.42E+05	2.41E-09

407 Note: Values correspond to a  $10^{-6}$  risk level for carcinogens and an HQ of 1 for non-carcinogens.

408 The table shows the HQ for PAHs for dermal contact for both adults and children. The values  
409 exceed one are shown in bold. They are for the char and the four soil samples collected  
410 nearest to the Tower. This corresponds to an increased risk of any adverse health effects  
411 from PAHs, but not from dioxins and furans. The table also shows the cancer risk to humans  
412 multiplied by a factor of  $10^6$ . Values exceeding  $1 \times 10^6$  indicate an increased cancer risk. These  
413 are also shown in bold in Table 5. The four soil samples (S1-S4) closest to the Tower indicate  
414 significantly increased cancer risk from dioxin and furans, as well as for PAHs, via dermal  
415 intake.

416 Soil samples, collected 6 months after the fire, show significant quantities of fire effluents. As  
417 soil samples were dried at 60 °C prior to analysis, reported levels of benzene, PAHs, and  
418 dioxins may be lower than actually presented. The distributions of benzene, PAHs and  
419 PCDD/Fs show very clearly that particulates/soot/char and fragments of fire debris were  
420 distributed within the vicinity of the Tower. Concentrations of PAHs, PCDD/Fs and benzene in  
421 the soils suggests that these particulates released toxic substances. Soil concentrations  
422 exceed guideline values within 150 m of the Tower. HCN was also observed in TGA-FTIR.  
423 Figure 6, shows the localisation of toxicant distribution following the fire. Results suggest that  
424 the major sources of contamination for the toxicants analysed in this study are particulates of  
425 diameter greater than 100 µm deposited up to 200 m away from the Tower and greater than  
426 10 µm within 2000 m distance from the Tower.



427

428 **Figure 6.** Concentrations of fire effluents in soil samples (Benzene and BaP in mg/kg,  
429 PCDD/Fs in µg/kg).

430 The elevated levels of dioxins and furans and PAHs found in soil samples is in stark contrast  
431 to the undetectable levels found during air monitoring by PHE (PHE 2018a). This is  
432 unsurprising since any gas phase PAHs or PCDD/Fs will have been dispersed prior to  
433 commenced of the PHE analysis (month after the fire) (PHE. 2018b).

434 The HCN evolution from the soil, mirrors the temperature range of release from ferri- and ferro-  
435 cyanides. This suggests that S1 to S4 were exposed to significant quantities of HCN,  
436 particularly as S5 –S7 show no such release. TGA-FTIR analysis showed release of MIC as  
437 a decomposition product of the fire debris and char samples. The yellow oil on the window  
438 blind has been previously characterised as a part polymerised product of isocyanates.  
439 Discovery of MIC deposition a volatile liquid (boiling point at 38 °C), on part of the window  
440 blind that was exposed to the outside air, is an obvious health concern particularly as it was  
441 found 17 months after the fire within a living space.

442 Analysis of the SVFs from the three insulation panels (PIR, PhF, SW) used on the Tower was  
443 compared to that of the SVFs found in soil, char and residue (see section 3.8). It was found  
444 that SVFs isolated from the soils are more likely to originate from PIR for S1 and PhF for S2.

445 The composition of the SVF from the fire debris corresponded to that of phenolic foam,  
446 whereas the SVF on the chars (Char2 and Char3) displayed close similarities to the SVF from  
447 the PIR foam.

448 Soon after the fire, there was little evidence of environmental, indoor or health surveillance to  
449 identify the types of fire effluents or populations at risk. The Grenfell Tower fire released both  
450 acute and chronic toxicants in the fire effluent which may have potential long-term adverse  
451 health effects on emergency responders, clean-up workers and local residents.

452 The data needs to be interpreted with caution as soil is a complex matrix which can vary  
453 significantly, even within a small area such as the Grenfell environments. A much more  
454 valuable study could have been undertaken in the immediate aftermath after the fire. The  
455 absorption and release of toxicants will depend both on their chemical nature and the  
456 characteristics of the soil. Sampling from better controlled environments such as plant pots,  
457 where a known potting compost has been used and the medium has been undisturbed since  
458 the fire, have potential to identify fire contaminants more reliably. In addition, indoor  
459 contaminants resulting from deposits within residents homes (dust) have greater potential for  
460 positive identification and establishing their relationship to any long-term health effects.

461 From earlier study on the fire behaviour of façade materials, it has been found that brominated  
462 flame retardants were not present in significant quantities on the exterior face of the building  
463 (McKenna *et al.* 2019). It is acknowledged that furniture and other products in the Tower may  
464 and will contain them, but less clear how much effluent would be released to the surrounding  
465 environment.

466 The presence of chlorinated PCDD/Fs and the presence of brominated flame retardants in  
467 furniture etc. suggests the likely presence of brominated and mixed brominated-chlorinated  
468 dioxins and furans. Significant quantities of chlorinated, brominated and mixed dioxins and  
469 furans were identified around the World Trade Centre (Landrigan *et al.* 2004). The presence  
470 of both PAHs and halogenated PCDD/Fs also strongly indicates the likely presence of

471 halogenated PAHs, polychlorinated and polybrominated biphenyls (Xu *et al.* 2018). None of  
472 these substances were analysed in this study. They have health risks associated with their  
473 presence and should be quantified in any follow-up study.

474 Any health effects, together with long-term fire exposure monitoring, should also be carried  
475 out and supervised by a multidisciplinary team with medical, environmental, fire and  
476 combustion toxicology expertise. Public agencies need to be adequately prepared to provide  
477 reliable guidance to the public on more appropriate means of exposure assessment, risk  
478 assessment, and preventive measures - in the event of a recurrence such as this tragic fire.

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